Appln No. 10/634,640 Amdt date March 3, 2006 Reply to Office action of November 3, 2005

Amendments to the Claims:

This listing of claims will replace all prior versions, and listings, of claims in the application:

Listing of Claims:

1. (Currently Amended) A method of decomposing an organic azide, comprising: allowing an organic azide to contact a catalyst that comprises a metal halide, main group halide, mixed metal-main group halide, or mixture thereof [[.]],

wherein the organic azide has the formula

$R-N_3$

where R is an organic group selected from the group consisting of alkyl, alkyl amino, nitrogen-containing heterocyclic-substituted alkyl, and alkyl amine substituted with at least one alkyl azide group.

- 2. (Cancelled)
- 3. (Original) A method as recited in claim 1, wherein the organic azide is 2-dimethylaminoethyl azide.
- 4. (Currently Amended) A method as recited in claim [[2,]] 1, wherein R is a nitrogen-containing heterocyclic-substituted alkyl group.
- 5. (Currently Amended) A method as recited in claim [[2,]] 1, wherein R is an alkyl amine substituted with at least one alkyl azide group.
- 6. (Original) A method as recited in claim 1, wherein the catalyst comprises a transition metal halide.

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- 7. (Original) A method as recited in claim 6, wherein the transition metal in the transition metal halide can have one or more formal oxidation states.
- 8. (Original) A method as recited in claim 6, wherein the transition metal in the transition metal halide is present in its highest formal oxidation state.
- 9. (Original) A method as recited in claim 1, wherein the catalyst comprises an iron halide or a mixture of iron halide and a second catalyst.
- 10. (Original) A method as recited in claim 1, wherein the catalyst comprises a transition metal chloride.
- 11. (Original) A method as recited in claim 10, wherein the transition metal chloride comprises iron (III) chloride, iron (II) chloride, or a combination of iron (III) chloride and iron (III) chloride.
- 12. (Original) A method as recited in claim 1, wherein the catalyst comprises an iron chloride in combination with a second catalyst.
- 13. (Original) A method as recited in claim 1, wherein the catalyst is dispersed on a support.
- 14. (Original) A method as recited in claim 13, wherein the support comprises a second organic halide decomposition catalyst.
 - 15. (Original) A method as recited in claim 1, wherein the catalyst is granular.
- 16. (Original) A method as recited in claim 15, wherein the catalyst is mixed with a different granular catalyst.
- 17. (Original) A method as recited in claim 15, wherein the catalyst is mixed with non-catalyst granules.

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- 18. (Currently Amended) A fuel source composition of matter comprising:
- (a) an organic azide having the formula

 $R-N_3$

where R is an organic group selected from the group consisting of alkyl, alkyl amino, nitrogencontaining heterocyclic-substituted alkyl, and alkyl amine substituted with at least one alkyl azide group; and

- (b) a catalyst capable of decomposing the organic azide, said catalyst comprising at least one metal halide, main group halide, mixed metal-main group halide, or mixture thereof.
- 19. (Currently Amended) A fuel source composition of matter as recited in claim 18, wherein the organic azide is 2-dimethylaminoethyl azide.
- 20. (Currently Amended) A fuel source composition of matter as recited in claim 18, wherein R is a nitrogen-containing heterocyclic-substituted alkyl group.
- 21. (Currently Amended) A fuel source composition of matter as recited in claim 18, wherein R is an alkyl amine substituted with at least one alkyl azide group.
- 22. (Currently Amended) A fuel source composition of matter as recited in claim 18, wherein the catalyst comprises a transition metal halide.
- 23. (Currently Amended) A <u>fuel source composition of matter</u> as recited in claim 22, wherein the transition metal in the transition metal halide can have one or more formal oxidation states.
- 24. (Currently Amended) A fuel source composition of matter as recited in claim 22, wherein the transition metal in the transition metal halide is present in its highest formal oxidation state.

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- 25. (Currently Amended) A fuel source composition of matter as recited in claim 18, wherein the catalyst comprises an iron halide or a mixture of iron halide and a second catalyst.
- 26. (Currently Amended) A fuel source composition of matter as recited in claim 18, wherein the catalyst comprises a transition metal chloride.
- 27. (Currently Amended) A fuel source composition of matter as recited in claim 26, wherein the transition metal chloride comprises iron (III) chloride, iron (III) chloride, or a combination of iron (III) chloride and iron (III) chloride.
- 28. (Currently Amended) A fuel source composition of matter as recited in claim 18, wherein the catalyst comprises an iron chloride in combination with a second catalyst.
- 29. (Currently Amended) A fuel source composition of matter as recited in claim 18, wherein the catalyst is dispersed on a support.
- 30. (Currently Amended) A fuel source composition of matter as recited in claim 29, wherein the support comprises a second organic halide decomposition catalyst.
- 31. (Currently Amended) A fuel source composition of matter as recited in claim 18, wherein the catalyst is granular.
- 32. (Currently Amended) A fuel source composition of matter as recited in claim 31, wherein the catalyst is mixed with a different granular catalyst.
- 33. (Currently Amended) A fuel source composition of matter as recited in claim 31, wherein the catalyst is mixed with non-catalyst granules.